

THE EFFECT OF THE SLURRY ADDITIVE NH₃RELIEF COMPARED TO SULFURIC ACID ON THE NITROGEN UPTAKE OF WINTER WHEAT

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Introduction

To prevent eutrophication of non-agricultural ecosystems, indirect N₂O emissions, particulate matter formation, and soil acidification, the agricultural NH₃ emissions have to be reduced [1][2][3]. By decreasing NH₃ emissions, the N fertilising effect and N use efficiency of the slurry might also be improved.

The product **NH₃relief** is a NH₃ sorbent based on carboxylic acid derivatives.

In a field trial, this sorbent was used as a slurry additive to improve the N fertilization effect of the treated slurry and was compared with the effect of sulfuric acid.

Materials and Methods

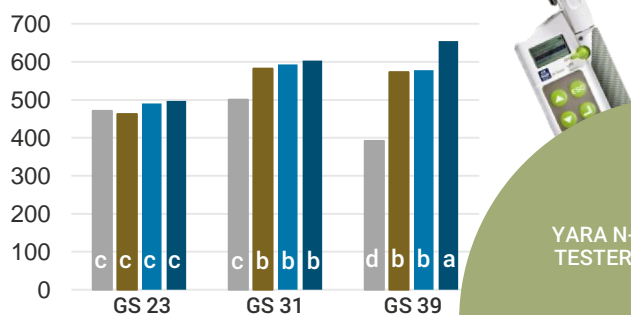
In the field trial (block design, 4 replicates), the winter wheat was fertilised two times using fattening pig slurry with a trailing hose system. The following 4 treatments were applied

1. Control without fertilizer,
2. Slurry without additive,
3. Slurry + NH₃relief (24 L m⁻³), and
4. Slurry + sulfuric acid (6 L m⁻³, pH 5.2).

At three growth stages **YARA N-Tester** measurements were recorded, plant samples taken and their **N concentrations** analysed. At harvest, **grain yield** and **N concentration** were determined.

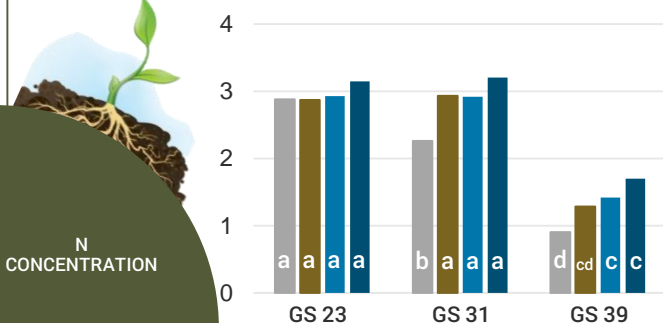
YARA N-Tester value

At three growth stages; n = 4; same letters -> no significant differences (Tukey contrasts, $\alpha = 0.05$)



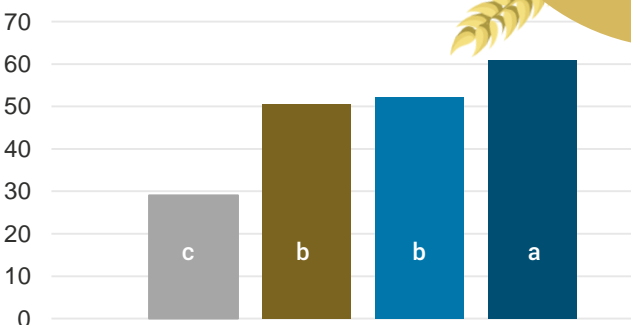
N concentration (% dry weight)

Whole-plant analysis at three growth stages; n = 4; same letters -> no significant differences (Tukey contrasts, $\alpha = 0.05$)



N removal (kg ha⁻¹)

Yield (dry weight) * grain N concentration (% dry weight); n = 4; same letters -> no significant differences (ANOVA; HSD ($\alpha = 0.05$))



Results

- Significantly lower YARA N-Tester value in response to slurry treated with NH₃relief compared to sulfuric acid at GS 39
- Lower (but not significantly) N concentration in the plant material for Slurry + NH₃relief compared to Slurry + sulfuric acid treatment at all growth stages
- Significantly lower N removal for Slurry + NH₃relief compared to Slurry + sulfuric acid treatment

Discussion

- Probably lower NH₃ emissions and thus higher NH₄⁺ supply due to the use of sulfuric acid and acidification of the slurry
 - Most likely reduced N availability of the Slurry + NH₃relief as NH₄⁺ might be bound to the carboxylic acid derivatives and thus not directly available to plants
- Further research on the availability of N bound to carboxylic acid derivatives is necessary.

REFERENCES

- [1] LELIEVELD et al. (2015): The contribution of out-door air pollution sources to premature mortality on a global scale. Nature 525, 367–371.
- [2] MOSIER, A. R. (2001): Exchange of gaseous nitrogen compounds between agricultural systems and the atmosphere. Plant and Soil 228, 17–27.
- [3] SPIRIG C. and NEFTEL, A. (2006): Ammoniakemissionen aus der Landwirtschaft und Feinstaub. Agrarforschung Schweiz 13, 392–397.

